

Cellulose-Based Materials: A Comprehensive Review for Sustainable Water Treatment Solutions

Deeksha Ranjan¹, Richa Choudhary^{2*}

¹Department of Applied Sciences and Humanities, Faculty of Engineering and Technology,
Rama University, Mandhana, Kanpur, U.P.-209217

Email: deeksha.rs.apc@itbhu.ac.in

^{2*}Department of Bio-Sciences, Division of Life Sciences, School of Basic and Applied Science,
Galgotias University, Greater NOIDA, U.P.-203201.

Email: richa.choudhary@galgotiasuniversity.edu.in

Abstract

Cellulose, a ubiquitous biopolymer derived from plant cell walls, finds extensive application in water treatment processes due to its remarkable properties. This abstract explores the diverse uses and benefits of cellulose in water treatment applications. Cellulose-based materials serve as effective adsorbents for removing various contaminants from water, including heavy metals, organic pollutants, and dyes, owing to their high surface area, porosity, and functional groups. Moreover, cellulose derivatives such as carboxymethyl cellulose and cellulose acetate enhance water treatment efficiency through membrane filtration processes, facilitating the separation of impurities from water streams. The eco-friendly nature of cellulose-based materials aligns with the growing demand for sustainable water treatment solutions, offering a renewable alternative to conventional synthetic polymers. Furthermore, the facile modification and functionalization of cellulose enable tailored designs for specific water treatment needs, enhancing performance and selectivity. Overall, cellulose-based materials exhibit immense potential in addressing water quality challenges, contributing to the development of efficient, cost-effective, and environmentally friendly water treatment technologies.

1. Cellulose based techniques for wastewater treatment:

Cellulose, one of the most found natural polymers on earth, consists of repeated units of β -D-glucopyranose linked through β -1,4-glycosidic linkage and derived from various wood sources. In its natural form cellulose has lesser applications. The functional groups present on cellulose surface can be modified with the help of physical and chemical methods and the derivatives are then employed for various applications including wastewater treatment. It may be modified by directly introducing inorganic nanoparticles as metal oxides, chelating agents or metal binders or by grafting of selected monomers into the cellulosic backbone as biopolymer chitosan (1-3). With modification and fabrication, cellulose can be employed for many commercial applications and products as membranes etc. Cellulose based nanomaterials can be prepared via various physical,

chemical, biological and mechanical methods. Nanoscale derivatives include cellulose nanocrystals, cellulose nanofibrils, cellulose nanofibrils (CNFs), and TEMPO (2,2,6,6-tetramethylpiperidine-1-oxyl radical)-mediated oxidized cellulose nanofibrils (TOCNF) (4, 5). The below section shows the possibility of cellulose and cellulose based materials in various wastewater treatment technologies.

1.1 Cellulose and its derivatives in precipitation:

Precipitation of water contaminants, such as metal ions and anionic species, is one of the oldest methods in water treatment. Precipitation of contaminants is carried out by adding various chemical precipitating reagents including cellulose. Cellulose has been reported to be used for the co-precipitation of various metal ions as Cd(II) (5). The precipitated pollutants are then removed via decantation or filtration. The filtration is carried out by filter paper or membrane. Commercial filter papers like Whatman® filter paper are made up of cellulose.

1.2 Cellulose and its derivatives for adsorption:

The main feature of cellulose is the presence of hydroxyl ions on its surface, which show affinity towards adsorption of heavy metals ions and organic ions. Raw agricultural residues which contain cellulose have low load capacity as well as selectivity as the -OH groups are involved in intermolecular hydrogen bonding (6). These hydrogen bonds can be destroyed by chemical modification of cellulose decreasing their crystalline structure and making the surface binding sites more efficient as -OH groups become more active to bind with other functional groups, finally enhancing the adsorptive property of cellulose. Various modified cellulose, grafted cellulose, cellulose beads and composites, nanocellulose, and cellulosic hydrogels have been employed for adsorption of organic and inorganic water pollutants (6) as shown in below table:

Table 3: Cellulose and derivatives used in adsorption of various water pollutants

| Adsorbent | Pollutant | Maximum adsorption capacity | Year | Reference |
|---|---|---|-------------|------------------|
| Neutral and variously modified cellulose nanofibrils | Multiple active pharmaceutical ingredients | 1.3 mg/g for MTF to 11.8 mg/g for DCF by unmodified LNPs | 2022 | 9 |
| Chemically modified cellulose fibers (PEI/CE) | Acid fuchsin (AF), Cu(II), perylene tetracarboxylate (PTC), and Zn(II) ions | AF - 562 mg/g Cu(II) - 552 mg/g PTC - 216 mg/g Zn(II) - 157 mg/g | 2022 | 10 |
| Sulfated Cellulose nanocrystals | Janus Green (JG) dye | 77 mg g ⁻¹ | 2022 | 11 |
| Spherical cellulose-based adsorbent (2-AMP resin) having 2-aminomethyl-pyridine | Ag(I) | - | 2020 | 12 |
| Composite nanofiber membrane of Deacetylated cellulose acetate and polydopamine | methylene blue (MB) | 88.2 mg/g | 2020 | 13 |

| | | | | |
|--|---|--|------|----|
| Cellulose isolated from straw was modulated using N(4)-morpholiniothiosemicarbazide | Ni(II) | 90 mg g ⁻¹ | 2020 | 14 |
| Thiosemicarbazide-modified cellulose | Cu(II) | 106.3829 mg g ⁻¹ | 2020 | 15 |
| Sodium Alginate/Hydroxypropyl Cellulose Beads | Pb (II) | 47.72 mg/g | 2020 | 16 |
| macrocyclic pyridone pentamer (MCP) with succinic acid anhydride modified microcrystalline cellulose (SA-MCC) | Ba(II), Pb(II), Cd(II), Mn(II), Cu(II), Co(II), Ni(II), Cr(III) | 75% of Ba ²⁺ and others less than 18% | 2019 | 17 |
| Thiosemicarbazide-modified cellulose | (Hg)II | 98.5% at a pH 5.0 | 2019 | 18 |
| Grafting amino-terminated hyperbranched polymer (NH ₂ -HBP) and beta-cyclodextrin (β-CD) onto cotton fibers | Congo red (CR) and methylene blue (MB) | 350.8 mg/g for CR and 102.7 mg/g for MB | 2019 | 19 |

| | | | | |
|---|--|--|------|----|
| Sulfonated cellulose | Cu ²⁺ | 8.2 mg-Cu ²⁺ /g | 2017 | 20 |
| Cellulose nanocrystal-alginate hydrogel beads | Continuous flow adsorption of methylene blue | 255.5 mg/g | 2015 | 21 |
| Cellulose nanocrystals and alginate (CNC-ALG hydrogel beads | methylene blue (MB) | 256.41 mg/g | 2015 | 22 |
| Carboxylated cellulose nanofibrils (CCNFs) | Pb(II) | 171.0mg/g | 2014 | 23 |
| Hydrogels based on cellulose-graft-poly (acrylic acid) copolymers (C-g-AA) | Cu(II) and Ni(II) | 182 and 200 mg/g for Cu(II) and Ni(II), respectively | 2016 | 24 |
| Cellulose formed with vinyl monomer glycidyl methacrylate by using ceric ammonium nitrate and functionalized with thiosemicarbazide | Hg(II), Cd(II), and acid fuchsin (AF) | - | 2013 | 13 |

| | | | | |
|--|------------------|---|------|----|
| Branched poly (ethylene imine) (PEI)-modified cellulose-based adsorbent (Cell-g-PGMA-PEI) | Cu(II) | 102 mg g ⁻¹ | 2012 | 25 |
| Cellulose wood pulp grafted with vinyl monomer glycidyl methacrylate (GMA) and functionalised with imidazole (cellulose-g-GMA-imidazole) | Pb(II), Ni(II) | 72 mg g ⁻¹ of Pb(II) and 45 mg g ⁻¹ of Ni(II) | 2006 | 26 |
| Carboxymethylcellulose (CMC) | Cu ²⁺ | - | 2004 | 27 |

Differently modified, grafted cellulose hydrogels, beads, and resins have been employed for the adsorption of water pollutants (28-32). O'Connell et al., 2008 (32) have listed a number of adsorbents prepared from modulation of cellulose for remediation of heavy metal ions. Hokkanen et al., 2016 (29) have reviewed various methods utilized to modulate cellulose based adsorbents to enhance their adsorptive property. A review published in 2021, by Akter et al. (30), concisely reviews the cellulose based hydrogels for wastewater treatment. The article reviews many heavy metal ions and organic dyes which have been adsorbed on the surface of cellulose based hydrogels. These various kinds of hydrogels based on cellulose have been reported to be superadsorbents and very durable, non-toxic, biodegradable and almost complete adsorption of heavy metal ions, dyes etc. from water as well as wastewater (33).

1.3 Cellulose and its derivatives for coagulation/ flocculation:

The main issue in addition of chemical coagulants/ flocculants as aluminium chloride and ferric chloride is the production of secondary pollutants. This can be resolved by employing biopolymers like cellulose as coagulant, which is eco-friendly and requires comparatively lesser steps. Cellulose based coagulants and flocculants are obtained by either by physically combining it with other material or by chemically modifying them or both (34,35). In research conducted by Mohamed Noor et al. in 2018 (36) cellulose was hybridized with a magnetite powder to form a flocculant to treat effluents from palm oil mill. Cellulose backbone has been modified with Ag₂O and TiO₂ and has been used for photodegradation (35). In a study carried out in 2010 (37) an anionic sodium carboxymethylated cellulose was formed from date palm rachis. It was combined with aluminium sulphate coagulant and was used as flocculant to remove turbidity while treatment of drinking water. Another study reports the use of anionized nanocellulose as flocculant combined with ferric sulphate coagulant in the treatment of municipal wastewater (38). There are reports of using sawdust derived cellulose nanocrystals as coagulant for Ni(II) and Cd(II) from water, showing adsorption power of 956.6 mg/g for Ni²⁺ and 2,207 mg/g for Cd²⁺ (39). It is also reported that modification with hexadecyltrimethylammonium bromide (HDTMA-Br) showed better performance in comparison to the commercially available coagulant R₂T₂ (40). A review published in 2021, by Fauzani et al. (41), mentions cellulose in natural flocculant applications. Another review article by Barrero-Fernández et al., in 2021 (42) does a bibliometric analysis on the application of cellulose-based materials used for flocculation.

1.4 Cellulose and its derivatives for membrane filtration:

In general, the physicochemical properties of membranes can be enhanced by application of chemical nanomaterials, which have been found to be toxic in nature. This has led to the use of low-cost, eco-friendly, and renewable cellulose or nanocellulose. Membranes for commercial purposes are fabricated by the use of cellulose like, polysulfone, polyethersulfone, polyvinylidene

fluoride, polypropylene (43, 44). In recent years nanocellulose, like cellulose nanocrystals, nanofibrils, bacterial nanocellulose have also been in application in development of different types of membranes, such as forward osmosis, membrane distillation, reverse osmosis, nanofiltration. Membranes based on nanocellulose have been found to be effective in desalination and thus commercial water/ wastewater treatment. Chemical modification of cellulose/ nanocellulose surface has also been found to be enhancing the surface properties and thus its affinity and reactivity towards effective removal of contaminants from water (Table 4) (45-47).

Table 4: Cellulose and derivatives used in membrane filtration for various water pollutants

| Membrane | Contaminants | Rejection rate/ Remarks | Year | Reference |
|--|---------------------------------|--|------|-----------|
| Application of Cellulose Nanocrystals in Nanofiltration | | | | |
| Nanocomposite membrane containing polydopamine-modified cellulose nanocrystals | Congo red (CR)/NaCl solution | Congo red rejection - 99.91% Salt permeation - 99.33% Mixture congo-red/NaCl selectivity - 98% | 2020 | 48 |
| Nanofiltration membrane from bamboo cellulose | NaCl | 36.11 % | 2021 | 49 |
| Polyamide nanofiltration membrane based on Cellulose | MgSO ₄ | Enhanced about 9.3% | 2020 | 50 |
| Thin film nanocomposite nanofiltration membrane incorporated with cellulose nanocrystal/silver nanocomposites into a polyamide layer | Na ₂ SO ₄ | 99.1% for Na ₂ SO ₄ . Also 99.4% reduction of <i>Escherichia coli</i> viability | 2020 | 51 |

| | | | | |
|--|---|--|------|----|
| Thin film composite nanofiltration membranes modified by cellulose nanocrystal sandwiched layer and a polydopamine layer | Congo red, Rose Bengal, sodium lignosulfonate and alkaline lignin | Excellent | 2020 | 52 |
| 2,2,6,6-tetramethylpiperidine-1-oxyl radical (TEMPO)-oxidized cellulose nanofibers (CNFs) incorporated into the polyamide layer of thin film composite (TFC) reverse osmosis (RO) membrane | NaCl | 96.2% | 2019 | 53 |
| Cellulose nanocrystals (CNCs) incorporated into polyamide (PA) layer to prepare thin film composite (TFC) nanofiltration membranes (CNC-TFC-Ms) | Na ₂ SO ₄ and MgSO ₄ | 98.0% and 97.5% for Na ₂ SO ₄ and MgSO ₄ , respectively | 2019 | 54 |
| Cellulose nanocrystals incorporated polyamide layer of thin film composite membrane | Na ₂ SO ₄ and MgSO ₄ | 98.7% and 98.8% respectively | 2018 | 55 |

| | | | | |
|---|--|---|------|----|
| Triple-layered composite nanofiltration membranes by interfacial polymerization of diamine and acyl chloride on cellulose nanocrystal interlayer supported by microporous substrate | Na ₂ SO ₄ , MgSO ₄ , MgCl ₂ , CaCl ₂ , NaCl | Na ₂ SO ₄ (97.7%) > MgSO ₄ (86%) > MgCl ₂ (15.5%) > CaCl ₂ (11%) > NaCl (6.5%) | 2017 | 56 |
| Application of Cellulose Nanocrystals in Reverse Osmosis | | | | |
| Carboxylated cellulose nanocrystal (CNCCR) incorporated active layer (AL-CNCCR) and supporting layer (SL-CNCCR) of TFC membrane | Na ₂ SO ₄ , MgSO ₄ , NaCl | AL-CNCCR rejection ratio increased to approximately 105.2%, and SL-CNCCR membranes showed 107.5% rejection ratio for sulphate ion. Also for sodium chloride; the increased value for chloride ion rejection of SL-CNCCR membrane (134.4%) and with AL-CNCCR membrane (108.7%) | 2021 | 57 |
| Sawdust-derived cellulose nanocrystals embedded in polyamide | sodium chloride, and calcium chloride | sodium chloride - 98.3 ± 0.8% calcium chloride - 97.1 ± 0.5% | 2020 | 58 |
| Cellulose nanocrystal membrane grafted with Poly(acryloyl hydrazide) | - | better water permeance, salt rejection, and organic fouling resistance, improved boron rejection | 2019 | 59 |

| | | | | |
|--|---------------------------|--|------|----|
| Reverse Osmosis nanocomposite membrane incorporated with 2,2,6,6-Tetramethylpiperidine-1-oxyl oxidized cellulose nanocrystals | Salt | 98.98 ± 0.41% | 2019 | 60 |
| Nanocomposite membrane incorporated with modified cellulose nanocrystals | Synthetic brackish water | 97.8%. Also, good fouling resistance properties for tested Bovine Serum Albumin | 2018 | 61 |
| Application of Cellulose Nanocrystals in Pervaporation (PV) | | | | |
| 1. Nanocomposite membranes based on poly(styrene)-block-poly(butadiene)-block-poly(styrene) (SBS) matrix and CNCs (SBS/CNC membranes) 2. decorated the CNCs with hydrophobic oleic acid moieties (OLA-CNCs) (SBS/OLA-CNC membranes) | Ethanol | 1. SBS/CNC membranes exhibit consistent reduction towards ethanol selectivity 2. SBS/OLA-CNC increased ethanol permeability | 2021 | 62 |
| CTA/CNCs membrane | NaCl hypersaline solution | 99.8% | 2021 | 63 |

| | | | | |
|--|---------------------------------|--|------|----|
| Functionalized cellulose nanocrystal incorporated TFN's polyamide thin layer | - | Good resistance for protein fouling and reduced water flux drop | 2020 | 64 |
| Cellulose triacetate/cellulose nanocrystals nanocomposite DMSO based pervaporation membranes | NaCl | 99.9% | 2020 | 65 |
| Cellulose triacetate/cellulose nanocrystals (CTA/CNCs) nanocomposite PV membrane. | NaCl | 99.9% | 2019 | 66 |
| Application of Cellulose Nanofibers in Nanofiltration | | | | |
| Nanofiltration Membranes on blended metal ion support | Dye and salt | Dye rejections for Congo red - 97.7%, Methyl blue - 97.1%, Eriochrome Black T - 95.0% Salt penetration rate for Na ₂ SO ₄ - 93.8%, MgSO ₄ - 95.1%, NaCl - 97.4%, MgCl ₂ - 98.1% | 2022 | 67 |
| Nanofiltration membrane composited with sulfated cellulose nanofibril | Na ₂ SO ₄ | Above 98% | 2022 | 67 |

| | | | | |
|--|---|--|------|----|
| Oxygen Plasma Treated composite membrane of Reduced Graphene Oxide/Cellulose Nanofiber | Acid Fuchsin, Rose Bengal, Brilliant Blue | Above 90% | 2021 | 68 |
| Application of Cellulose Nanofibers in Ultrafiltration | | | | |
| High-lignin unbleached neutral sulfite's cellulose nanofibers for ultrafiltration | - | water flux efficiency was 96% higher | 2020 | 69 |
| Surface modification via interfacial polymerization of ultrafine cellulose nanofiber (UCN) membranes | MgCl ₂ , MgSO ₄ , NaCl, Na ₂ SO ₄ | MgCl ₂ —89.7%, MgSO ₄ —65.3%, NaCl—43.6%, and Na ₂ SO ₄ (39.1%) | 2017 | 70 |
| Application of Cellulose Nanofibers in Reverse Osmosis | | | | |
| Reverse osmosis membranes of aromatic polyamide (PA) reinforced with a crystalline cellulose nanofiber (CNF) | Chloride ions | higher hydrophilicity and higher water permeability, enhanced antifouling performance and improved chlorine resistance | 2020 | 71 |
| Composite reverse osmosis membrane of polyamide layer incorporated with 2,2,6,6-tetramethylpiperidine-1-oxyl | NaCl | 96.2% | 2019 | 72 |

| | | | | |
|---|--|---|------|----|
| radical (TEMPO)-oxidized cellulose nanofibers | | | | |
| Oxidized cellulose nanofibers membrane modified with Cu-terpyridine | Cr(VI) | 80% | 2018 | 73 |
| Cellulose Acetate/Cellulose Nanofiber Membranes | whey, strawberry juice and raspberry juice | Reduced turbidity and solid content | 2017 | 74 |
| Application of Bacterial Nano Cellulose (BNC) in Membrane distillation | | | | |
| Polydopamine particles and bacterial nanocellulose advanced membrane | salt | >99.9% | 2021 | 75 |
| Unsupported bacterial nanocellulose aerogel membranes | - | higher intrinsic membrane permeability and thermal efficiency | 2016 | 76 |
| Application of Bacterial Nano Cellulose (BNC) in Ultrafiltration | | | | |

| | | | | |
|--|--|--|------|----|
| Membrane composed of polydopamine (PDA) particles and bacterial nanocellulose (BNC) | Pb, Cd, Rhodamine 6G (R6G), methylene blue (MB), and methyl orange (MO) | easy to fabricate, highly scalable, chemically stable, mechanically robust, and reusable | 2019 | 77 |
| Ultrafiltration membrane of reduced graphene oxide and bacterial nanocellulose | - | two-fold increment of water flux | 2019 | 78 |
| Membrane based on bacterial nanocellulose (BNC) loaded with graphene oxide (GO) and palladium (Pd) nanoparticles | Methyl orange, Multiple contaminants (4-nitrophenol, methylene blue, and rhodamine 6G) | highly efficient methylene orange (MO) degradation during filtration (up to 99.3%) | 2018 | 79 |

Conclusion

In conclusion, this review highlights the significant role of cellulose-based materials in advancing sustainable water treatment solutions. Through an exploration of various applications and properties, it is evident that cellulose offers versatile benefits for addressing water quality challenges. From its ability to adsorb contaminants to its effectiveness in membrane filtration processes, cellulose demonstrates promise as a renewable and eco-friendly alternative to traditional materials. The review underscores the importance of cellulose derivatives, such as carboxymethyl cellulose and cellulose acetate, in enhancing water treatment efficiency and selectivity. Moreover, the facile modification and functionalization of cellulose enable tailored designs to meet specific water treatment needs. As society

increasingly prioritizes environmental sustainability, the use of cellulose in water treatment aligns with these objectives, offering a renewable and biodegradable solution. Moving forward, further research and development in cellulose-based materials hold potential for innovation in water treatment technologies, paving the way for more efficient, cost-effective, and environmentally friendly solutions to ensure access to clean water for all.

References

1. Saravanan, R., Ravikumar, L. (2015). The use of new chemically modified cellulose for heavy metal ion adsorption and antimicrobial activities. *J. Water Resour. Prot.*, 7, 530–545.
2. Singh, A. S., Guleria, A. Chemical modification of cellulosic biopolymer and its use in removal of heavy metal ions from wastewater. (2014), *Int J Biol Macromol* 67:409–417.
3. Kumar, R., Sharma, R. K., & Singh, A. P. (2019). Grafting of cellulose with N-isopropylacrylamide and glycidyl methacrylate for efficient removal of Ni(II), Cu(II) and Pd(II) ions from aqueous solution. *Separation and Purification Technology*, 219, 249-259. <https://doi.org/https://doi.org/10.1016/j.seppur.2019.03.035>
4. Xie, H., Du, H., Yang, X., & Chuanling, S. (2018). Recent Strategies in Preparation of Cellulose Nanocrystals and Cellulose Nanofibrils Derived from Raw Cellulose Materials. *International Journal of Polymer Science*, 2018, 1-25. <https://doi.org/10.1155/2018/7923068>
5. Sharma, P., Chattopadhyay, A., Sharma, S., Geng, L.-H., Amiralian, N., Martin, D., & Hsiao, B. (2018). Nanocellulose from Spinifex as an Effective Adsorbent to Remove Cadmium(II) from Water. *ACS Sustainable Chemistry & Engineering*, 6. <https://doi.org/10.1021/acssuschemeng.7b03473>
6. Pan, B., Pan, B., Zhang, W., Lv, L., Zhang, Q., & Zheng, S. (2009). Development of polymeric and polymer-based hybrid adsorbents for pollutants removal from waters. *Chemical Engineering Journal*, 151(1), 19-29. <https://doi.org/https://doi.org/10.1016/j.cej.2009.02.036>
7. Melissa B. Agustin, Kirsi S. Mikkonen, Marianna Kemell, Panu Lahtinen and Mari Lehtonen, Systematic investigation of the adsorption potential of lignin- and cellulose-based nanomaterials towards pharmaceuticals, *Environ. Sci.: Nano*, 2022,9, 2006-2019
8. Batool, A., & Valiyaveetil, S. (2022). Sequential Removal of Oppositely Charged Multiple Compounds from Water Using Surface-Modified Cellulose. *Industrial & Engineering Chemistry Research*, 61(1), 716-726. <https://doi.org/10.1021/acs.iecr.1c03847>
9. Aggarwal, R., Garg, A.K., Saini, D., Sonkar, S.K., Sonker, A.K., Westman, G. (2022). Cellulose Nanocrystals Derived from Microcrystalline Cellulose for Selective Removal of Janus Green Azo Dye. *Ind. Eng. Chem. Res.* 2023, 62, 1, 649–659.
10. Yang, X., Dong, Z., Zhang, M., Du, J., & Zhao, L. (2020). Selective Recovery of Ag(I) Using a Cellulose-Based Adsorbent in High Saline Solution. *Journal of Chemical & Engineering Data*, 65. <https://doi.org/10.1021/acs.jced.9b01107>
11. Cheng, J., Zhan, C., Wu, J., Cui, Z., Si, J., Wang, Q., . . . Turng, L. S. (2020), Highly Efficient Removal of Methylene Blue Dye from an Aqueous Solution Using Cellulose

- Acetate Nanofibrous Membranes Modified by Polydopamine. 5, 10, 5389–5400. (2470-1343 (Electronic)).
12. Vu, H.T., Phan, M.T.D., Tran, U.T.T., Nguyen, G.D., Duong, V.B., Tran, D.B. (2020). N(4)-Morpholiniothiosemicarbazide-Modified Cellulose: Synthesis, Structure, Kinetics, Thermodynamics, and Ni(II) Removal Studies, *ACS Omega*, 5 (25), 15229-15239.
 13. Nguyen, T.A., Tran, D., Le, H., Nguyen, Q., & Pham, V. (2020). Thiosemicarbazone-Modified Cellulose: Synthesis, Characterization, and Adsorption Studies on Cu(II) Removal. *ACS Omega*, 5 (24), 14481-14493.
 14. Guerrero, R., Acibar, C., Alarde, C. M., Maslog, J., & Pacilan, C. J. (2020). Evaluation of Pb (II) Removal from Water Using Sodium Alginate/Hydroxypropyl Cellulose Beads. *E3S Web of Conferences*, 148, 02002. <https://doi.org/10.1051/e3sconf/202014802002>
 15. Zhou, C., Ni, J., Zhang, D., & Sun, C. Cellulosic adsorbent functionalized with macrocyclic pyridone pentamer for selectively removing metal cations from aqueous solutions. (1879-1344 (Electronic)).
 16. Jiang, J., Wang, X. (2019). Adsorption of Hg(II) ions from aqueous solution by thiosemicarbazide-modified cellulose adsorbent. *BioRes.* 14(2), 4670-4695.
 17. Yue, X., Jiwei, H., Jiang, F., & Chen, Y. (2019). Synthesis and characterization of cellulose-based adsorbent for removal of anionic and cationic dyes. *Journal of Engineered Fibers and Fabrics*, 14, 155892501982819. <https://doi.org/10.1177/1558925019828194>
 18. Parlak, E., & Arar, Ö. (2017). Removal of copper (Cu 2+) from water by sulfonated cellulose. *Journal of Dispersion Science and Technology*, 39, 1-6. <https://doi.org/10.1080/01932691.2017.1405818>
 19. Mohammed, N., Grishkewich, N., Waeijen, H. A., Berry, R. M., & Tam, K. C. (2016). Continuous flow adsorption of methylene blue by cellulose nanocrystal-alginate hydrogel beads in fixed bed columns. *Carbohydrate Polymers*, 136, 1194-1202. <https://doi.org/https://doi.org/10.1016/j.carbpol.2015.09.099>
 20. Mohammed, N., Grishkewich, N., Berry, R., & Tam, K. (2015). Cellulose nanocrystal–alginate hydrogel beads as novel adsorbents for organic dyes in aqueous solutions. *Cellulose*, 22. <https://doi.org/10.1007/s10570-015-0747-3>
 21. Zhou, Y., Fu, S., Zhang, L., Zhan, H., & Levit, M. V. (2014). Use of carboxylated cellulose nanofibrils-filled magnetic chitosan hydrogel beads as adsorbents for Pb(II). *Carbohydrate Polymers*, 101, 75-82. <https://doi.org/https://doi.org/10.1016/j.carbpol.2013.08.055>
 22. Yang, S., Fu, S., Liu, J., Zhou, Y., Li, X., & Liu, Q. (2016). Adsorption of Hydrogels Based on Cellulose for Cu(II) and Ni(II): Behaviors and Mechanisms. *Journal of Macromolecular Science, Part B*, 55:7, 722-731.
 23. Tang, Y., Ma, Q., Luo, Y., Zhai, L., Che, Y., & Meng, F. (2013). Improved synthesis of a branched poly(ethylene imine)-modified cellulose-based adsorbent for removal and recovery of Cu(II) from aqueous solution. *Journal of Applied Polymer Science*, 129(4), 1799-1805. <https://doi.org/https://doi.org/10.1002/app.38878>

24. O'Connell, D. W., Birkinshaw, C., & O'Dwyer, T. F. (2006). Design Of A Novel Cellulose-based Adsorbent For Use In Heavy Metal Recovery From Aqueous Waste Streams. *WIT Transactions on Ecology and the Environment*, 95.
25. Hara, K., Iida M Fau - Yano, K., Yano K Fau - Nishida, T., & Nishida, T. Metal ion absorption of carboxymethylcellulose gel formed by gamma-ray irradiation. For the environmental purification. (0927-7765 (Print)).
26. Nakakubo, K., Hasegawa, H., Ito, M., Yamazaki, K., Miyaguchi, M., Biswas, F. B., . . . Maeda, K. (2019). Dithiocarbamate-modified cellulose resins: A novel adsorbent for selective removal of arsenite from aqueous media. *Journal of Hazardous Materials*, 380, 120816. <https://doi.org/https://doi.org/10.1016/j.jhazmat.2019.120816>.
27. Kumar, R., & Sharma, R. K. (2019). Synthesis and characterization of cellulose based adsorbents for removal of Ni(II), Cu(II) and Pb(II) ions from aqueous solutions. *Reactive and Functional Polymers*, 140, 82-92. <https://doi.org/https://doi.org/10.1016/j.reactfunctpolym.2019.04.014>
28. Chen, Y., Long, Y., Li, Q., Chen, X., & Xu, X. (2019). Synthesis of high-performance sodium carboxymethyl cellulose-based adsorbent for effective removal of methylene blue and Pb (II). *International Journal of Biological Macromolecules*, 126, 107-117. <https://doi.org/https://doi.org/10.1016/j.ijbiomac.2018.12.119>
29. Hokkanen, S., Bhatnagar, A., & Sillanpää, M. (2016). A review on modification methods to cellulose-based adsorbents to improve adsorption capacity. *Water Research*, 91, 156-173. <https://doi.org/https://doi.org/10.1016/j.watres.2016.01.008>
30. Akter, M., Bhattacharjee, M., Dhar, A.K., Rahman, F.B.A., Haque, S., Rashid, T.U., Kabir, S.M.F. (2021). Cellulose-Based Hydrogels for Wastewater Treatment: A Concise Review. *Gels*, 7(1), 30.
31. Zhou, Y., Hu, X., Zhang, M., Zhuo, X., & Niu, J. (2013). Preparation and Characterization of Modified Cellulose for Adsorption of Cd(II), Hg(II), and Acid Fuchsin from Aqueous Solutions. *Industrial & Engineering Chemistry Research*, 52(2), 876-884. <https://doi.org/10.1021/ie301742h>
32. O'Connell, D. W., Birkinshaw, C., & O'Dwyer, T. F. (2008). Heavy metal adsorbents prepared from the modification of cellulose: A review. *Bioresource Technology*, 99(15), 6709-6724. <https://doi.org/https://doi.org/10.1016/j.biortech.2008.01.036>
33. Tu, H., Yu, Y., Chen, J., Shi, X., Zhou, J., Deng, H., & Du, Y. (2017). Highly cost-effective and high-strength hydrogels as dye adsorbents from natural polymers: chitosan and cellulose. *Polymer Chemistry*, 8(19), 2913-2921. <https://doi.org/https://doi.org/10.1039/c7py00223h>
34. Lee, K. E., Morad, N., Teng, T. T., & Poh, B. T. (2012). Development, characterization and the application of hybrid materials in coagulation/flocculation of wastewater: A review. *Chemical Engineering Journal*, 203, 370-386. <https://doi.org/https://doi.org/10.1016/j.cej.2012.06.109>

35. Koshani, R., Tavakolian, M., & van de Ven, T. A.-O. Cellulose-based dispersants and flocculants. (2050-7518 (Electronic)).
36. Mohamed Noor, M., Ngadi, N., & Syie Luing, W. (2018a). Synthesis of Magnetic Cellulose as Flocculant for Pre- Treatment of Anaerobically Treated Palm Oil Mill Effluent. *Chemical Engineering Transactions*, 63, 589-594. <https://doi.org/10.3303/CET1863099>
37. Khiari, R., Dridi-Dhaouadi S Fau - Aguir, C., Aguir C Fau - Mhenni, M. F., & Mhenni, M. F. (2010), Experimental evaluation of eco-friendly flocculants prepared from date palm rachis. *J Environ Sci* 22:1539–1543.
38. Suopajarvi, T., Liimatainen, H., Hormi, O., & Niinimäki, J. (2013). Coagulation– flocculation treatment of municipal wastewater based on anionized nanocelluloses. *Chemical Engineering Journal*, 231, 59-67. <https://doi.org/https://doi.org/10.1016/j.cej.2013.07.010>
39. Oyewo, O. A., Mutesse, B., Leswif, T. Y., and Onyango, M. S. (2019). Highly Efficient Removal of Nickel and Cadmium from Water Using Sawdust-Derived Cellulose Nanocrystals. *J. Environ. Chem. Eng.* 7(4), 103251. doi:10.1016/j.jece.2019.103251
40. Nkalane, A., Oyewo, O. A., Leswif, T., and Onyango, M. S. (2019). Application of Coagulant Obtained through Charge Reversal of Sawdust-Derived Cellulose Nanocrystals in the Enhancement of Water Turbidity Removal. *Mater. Res. Express* 6, 105060.
41. Fauzani, D., Notodarmojo, S., Handajani, M., Helmy, Q., & Kardiansyah, T. (2021). Cellulose in natural flocculant applications: A review. *Journal of Physics: Conference Series*, 2047, 012030. <https://doi.org/10.1088/1742-6596/2047/1/012030>
42. Barrero-Fernández, A., Aguado, R., Moral, A., Brindley, C., & Ballesteros, M. (2021). Applications of cellulose-based agents for flocculation processes: a bibliometric analysis. *Cellulose*, 28, 9857–9871. <https://doi.org/10.1007/s10570-021-04122-z>
43. Goethem, C., Verbeke, R., Pfanmöller, M., Koschine, T., Dickmann, M., Stimpel-Lindner, T., . . . Vankelecom, I. (2018). The role of MOFs in Thin-Film Nanocomposite (TFN) membranes. *J. Memb. Sci.*, 563, 938–948. <https://doi.org/10.1016/j.memsci.2018.06.040>
44. Mavukkandy, M. O., McBride, S. A., Warsinger, D. M., Dizge, N., Hasan, S. W., & Arafat, H. A. (2020). Thin film deposition techniques for polymeric membranes– A review. *J. Memb. Sci.*, 610, 118258. <https://doi.org/https://doi.org/10.1016/j.memsci.2020.118258>
45. Tan, H.-F., Ooi, B. S., & Leo, C. P. (2020). Future perspectives of nanocellulose-based membrane for water treatment. *Journal of Water Process Engineering*, 37, 101502. <https://doi.org/https://doi.org/10.1016/j.jwpe.2020.101502>
46. George, J., Sabapathi, S. N. (2015), Cellulose nanocrystals: synthesis, functional properties, and applications, *Nanotechnol. Sci. Appl.*, 8, 45–54.
47. Abol-Fotouh, D., Hassan, M.A., Shokry, H., Roig, A., Azab, M.S., Kashyout, A.E.H.B. (2020). Bacterial Nanocellulose from Agro-Industrial Wastes: Low-Cost and Enhanced Production by *Komagataeibacter Saccharivorans* MD1. *Sci. Rep.* 10, 1–14.

48. Yang, L., Liu, X., Zhang, X., Chen, T., Ye, Z., & Rahaman, M. S. (2022). High performance nanocomposite nanofiltration membranes with polydopamine-modified cellulose nanocrystals for efficient dye/salt separation. *Desalination*, 521, 115385. <https://doi.org/10.1016/j.desal.2021.115385>
49. Li, S., Wang, D., Xiao, H., Zhang, H., Cao, S., Chen, L., . . . Huang, L. (2021). Ultra-low pressure cellulose-based nanofiltration membrane fabricated on layer-by-layer assembly for efficient sodium chloride removal. *Carbohydrate Polymers*, 255, 117352. <https://doi.org/10.1016/j.carbpol.2020.117352>
50. Wang, D.; Yuan, H.; Chen, Y.; Ni, Y.; Huang, L.; Mondal, A.K.; Lin, S.; Huang, F.; Zhang, H. (2020), A Cellulose-Based Nanofiltration Membrane with a Stable Three-Layer Structure for the Treatment of Drinking Water. *Cellulose*, 27, 8237–8253.
51. Xu, C., Chen, W., Gao, H., Xie, X., & Chen, Y. (2020). Cellulose Nanocrystal/Silver (CNC/Ag) Thin-film Nanocomposite Nanofiltration Membranes with Multifunctional Properties. *Environmental Science: Nano*, 7, 803–816. <https://doi.org/10.1039/C9EN01367A>
52. Bai, L., Ding, J., Wang, H., Ren, N., Li, G., & Liang, H. (2020), High-performance nanofiltration membranes with a sandwiched layer and a surface layer for desalination and environmental pollutant removal, 743, 140766. (1879-1026 (Electronic)).
53. Liu, S., Low, Z.-X., Hegab, H. M., Xie, Z., Ou, R., Yang, G., . . . Wang, H. (2019). Enhancement of desalination performance of thin-film nanocomposite membrane by cellulose nanofibers. *Journal of Membrane Science*, 592, 117363. <https://doi.org/10.1016/j.memsci.2019.117363>
54. Bai, L., Liu, Y., Ding, A., Ren, N., Li, G., & Liang, H. (2019). Fabrication and characterization of thin-film composite (TFC) nanofiltration membranes incorporated with cellulose nanocrystals (CNCs) for enhanced desalination performance and dye removal. *Chemical Engineering Journal*, 358, 1519-1528. <https://doi.org/10.1016/j.cej.2018.10.147>
55. Bai, L., Liu, Y., Bossa, N., Ding, A., Ren, N., Li, G., Liang, H., Wiesner, M.R. (2018), Incorporation of Cellulose Nanocrystals (CNCs) into the Polyamide Layer of Thin-Film Composite (TFC) Nanofiltration Membranes for Enhanced Separation Performance and Antifouling Properties. *Environ. Sci. Technol.*, 52, 11178–11187.
56. Wang, J.-J., Yang, H.-C., Wu, M., Zhang, X., & Xu, Z.-K. (2017a). Nanofiltration Membranes with Cellulose Nanocrystals as an Interlayer for Unprecedented Performance. *J. Mater. Chem. A*, 5. <https://doi.org/10.1039/C7TA00501F>
57. Liu, Y., Bai, L., Zhu, X., Xu, D., Li, G., Liang, H., & Wiesner, M. R. (2021). The role of carboxylated cellulose nanocrystals placement in the performance of thin-film composite (TFC) membrane. *Journal of Membrane Science*, 617, 118581. <https://doi.org/10.1016/j.memsci.2020.118581>
58. Adeniyi, A., Gonzalez-Ortiz, D., Pochat-Bohatier, C., Oyewo, O., Sithole, B., & Onyango, M. (2020). Incorporation of Cellulose Nanocrystals (CNC) derived from sawdust into

- polyamide thin-film composite membranes for enhanced water recovery. *Alexandria Engineering Journal*, 59(6), 4201-4210.
<https://doi.org/https://doi.org/10.1016/j.aej.2020.07.025>
59. Park, C., Jeon, S., Park, S.-H., Shin, M., Park, M., Lee, S., & Lee, J.-H. (2019). Cellulose Nanocrystal-assembled Reverse Osmosis Membranes with High Rejection Performance and Excellent Antifouling. *Journal of Materials Chemistry A.*, 7, 3992–4001.
<https://doi.org/10.1039/C8TA10932J>
60. Smith, E. A.-O., Hendren, K. A.-O. X., Haag, J. V. t., Foster, E. A.-O., & Martin, S. M. Functionalized Cellulose Nanocrystal Nanocomposite Membranes with Controlled Interfacial Transport for Improved Reverse Osmosis Performance. LID - 10.3390/nano9010125 [doi] LID - 125. (2019-4991 (Print)).
61. Asempour, F., Emadzadeh, D., Matsuura, T., & Kruczek, B. (2018). Synthesis and characterization of novel Cellulose Nanocrystals-based Thin Film Nanocomposite membranes for reverse osmosis applications. *Desalination*, 439, 179-187.
<https://doi.org/https://doi.org/10.1016/j.desal.2018.04.009>
62. Kamtsikakis, A., Delepierre, G., & Weder, C. (2021). Cellulose nanocrystals as a tunable nanomaterial for pervaporation membranes with asymmetric transport properties. *Journal of Membrane Science*, 635, 119473.
<https://doi.org/https://doi.org/10.1016/j.memsci.2021.119473>
63. Prihatiningtyas, I., Hartanto, Y., & Van der Bruggen, B. (2021). Ultra-high flux alkali-treated cellulose triacetate/cellulose nanocrystal nanocomposite membrane for pervaporation desalination. *Chemical Engineering Science*, 231, 116276.
<https://doi.org/https://doi.org/10.1016/j.ces.2020.116276>
64. Rahimi-Kashkouli, Y., Rahbari Sisakht, M., & Ghadami Jadval Ghadam, A. (2020). Thin film nanocomposite nanofiltration membrane incorporated with cellulose nanocrystals with superior anti-organic fouling affinity. *Environmental Science: Water Research & Technology*, 6, 715-723. <https://doi.org/10.1039/C9EW00963A>
65. Prihatiningtyas, I., Li, Y., Hartanto, Y., Vananroye, A., Coenen, N., & Van der Bruggen, B. (2020). Effect of solvent on the morphology and performance of cellulose triacetate membrane/cellulose nanocrystal nanocomposite pervaporation desalination membranes. *Chemical Engineering Journal*, 388, 124216.
<https://doi.org/https://doi.org/10.1016/j.cej.2020.124216>
66. Prihatiningtyas, I., Volodin, A., & Van der Bruggen, B. (2019). 110th Anniversary: Cellulose nanocrystals (CNCs) as organic nanofillers for cellulose triacetate membranes used for desalination by pervaporation. *Ind. Eng. Chem. Res.*, 58, 14340–14349.
<https://doi.org/10.1021/acs.iecr.9b02106>
67. Fang, X., Wei, S., Liu, S., Li, R., Zhang, Z., Liu, Y., . . . Li, F. (2022), Metal-Coordinated Nanofiltration Membranes Constructed on Metal Ions Blended Support toward Enhanced Dye/Salt Separation and Antifouling Performances. *Membranes*, 12, 340.

68. Mohammed, S., Hegab, H. M., Ou, R., Liu, S., Ma, H., Chen, X., . . . Wang, H. (2021). Effect of oxygen plasma treatment on the nanofiltration performance of reduced graphene oxide/cellulose nanofiber composite membranes. *Green Chemical Engineering*, 2(1), 122-131. <https://doi.org/10.1016/j.gce.2020.12.001>
69. Hassan, M.L., Fadel, S.M., Abouzeid, R.E., Abou Elseoud, W.S., Hassan, E.A., Berglund, L., Oksman, K. (2020), Water Purification Ultrafiltration Membranes Using Nanofibers from Unbleached and Bleached Rice Straw. *Sci. Rep.*, 10, 1–9.
70. Soyekwo, F., Zhang, Q., Gao, R., Qu, Y., Lin, C., Huang, X., . . . Liu, Q. (2017). Cellulose nanofiber intermediary to fabricate highly-permeable ultrathin nanofiltration membranes for fast water purification. *Journal of Membrane Science*, 524, 174-185. <https://doi.org/10.1016/j.memsci.2016.11.019>
71. Cruz-Silva, R., Izu K Fau - Maeda, J., Maeda J Fau - Saito, S., Saito S Fau - Morelos-Gomez, A., Morelos-Gomez A Fau - Aguilar, C., Aguilar C Fau - Takizawa, Y., . . . Endo, M. (2020), Nanocomposite desalination membranes made of aromatic polyamide with cellulose nanofibers: synthesis, performance, and water diffusion study. *Nanoscale*, 12, 19628–19637.
72. Liu, S., Low, Z.-X., Hegab, H. M., Xie, Z., Ou, R., Yang, G., . . . Wang, H. (2019). Enhancement of desalination performance of thin-film nanocomposite membrane by cellulose nanofibers. *Journal of Membrane Science*, 592, 117363. <https://doi.org/10.1016/j.memsci.2019.117363>
73. Hassan, M.; Hassan, E.; Fadel, S.M.; Abou-Zeid, R.E.; Berglund, L.; Oksman, K.(2018), Metallo-Terpyridine-Modified Cellulose Nanofiber Membranes for Papermaking Wastewater Purification. *J. Inorg. Organomet. Polym. Mater.*, 28, 439–447.
74. Battirola, L., Andrade, P., Marson, G., Hubinger, M., & Gonçalves, M. (2017). Cellulose acetate/cellulose nanofiber membranes for whey and fruit juice microfiltration. *Cellulose*, 24, 5593–5604. <https://doi.org/10.1007/s10570-017-1510-8>
75. Wu, X., Cao, S., Ghim, D., Jiang, Q., Singamaneni, S., & Jun, Y.-S. (2021). A thermally engineered polydopamine and bacterial nanocellulose bilayer membrane for photothermal membrane distillation with bactericidal capability. *Nano Energy*, 79, 105353. <https://doi.org/10.1016/j.nanoen.2020.105353>
76. Leitch, M., Li, C., Ikkala, O., Mauter, M., & Lowry, G. (2016). Bacterial Nanocellulose Aerogel Membranes: Novel High-Porosity Materials for Membrane Distillation. *Environmental Science & Technology Letters*, 3, 85–91. <https://doi.org/10.1021/acs.estlett.6b00030>
77. Gholami Derami, H., Jiang, Q., Ghim, D., Cao, S., Chandar, Y.J., Morrissey, J.J., Jun, Y.S., Singamaneni, S. (2019), A Robust and Scalable Polydopamine/Bacterial Nanocellulose Hybrid Membrane for Efficient Wastewater Treatment. *ACS Appl. Nano Mater.*, 2, 1092–1101.
78. Jiang, Q.; Ghim, D.; Cao, S.; Tadepalli, S.; Liu, K.K.; Kwon, H.; Luan, J.; Min, Y.; Jun, Y.S.; Singamaneni, S. Photothermally Active Reduced Graphene Oxide/Bacterial

Nanocellulose Composites as Biofouling-Resistant Ultrafiltration Membranes. *Environ. Sci. Technol.* **2019**, *53*, 412–421.

- 79.** Xu, T., Jiang, Q., Ghim, D., Liu, K.K., Sun, H., Derami, H.G., Wang, Z., Tadepalli, S., Jun, Y.S., Zhang, Q., et al. (2018), Catalytically Active Bacterial Nanocellulose-Based Ultrafiltration Membrane. *Small*, *14*, 1–8.